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Short Communication

Determination of hypericin and pseudohypericin by thinlayer chromatography—densitometry

Th. Kartnig and I. Göbel

Institute of Pharmacognosy, Karl-Franzens-University, Universitätsplatz 4/I, A-8010 Graz (Austria)

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ABSTRACT

The accuracy and reproducibility of the determination of hypericin and pseudohypericin by measuring the emitted fluorescence in situ on thin-layer chromatography plates can be improved significantly by using appropriate dipping reagents. The effects of pyridine, polyethylene glycol 4000, paraffins, silicone oil and Triton X-100 in various solvents and combinations on the intensity and stability of emitted fluorescence are described. Using the test conditions established by this investigation, hypericin can be quantified exactly within the range 5–50 ng and pseudohypericin within the range 20–200 ng. The relative standard deviation for the quantification under the given conditions is less than 1.5% for both substances.

INTRODUCTION

Thin-layer chromatography (TLC)—densitometry is still an appropriate method of determining hypericin and pseudohypericin in plant and plant tissue extracts as well as in pharmaceutical formulations [1]. Several TLC systems for the separation of hypericin and pseudohypericin are described [1–5].

The exact and reproducible quantification of TLC-separated hypericins by measuring the emitted fluorescence in situ depends on intensifying and stabilizing the fluorescence. This may be achieved by spraying or dipping the TLC plate with appropriate reagents after development. A review on the use of fluorescence enhancement reagents in TLC is given in ref. 6. In systematic investigations we tested sub-

stances reported to increase the intensity of fluorescence of various plant constituents [2,3,7–12]. Finally, we improved the conditions of quantification significantly so that an exact determination of hypericin within the range 5–50 ng and of pseudohypericin within the range 20–200 ng with a relative standard deviation of less than 1.5% was achieved.

EXPERIMENTAL

Chemicals

Hypericin was obtained from Roth (Karlsruhe, Germany, product No. 7929 and was used as follows: 1 mg in 100 ml of methanol (solution A), 1 ml of solution A diluted with 3 ml of methanol (solution B) (2 μ l of solution B = 5 ng of hypericin). Pseudohypericin was obtained from Hämosan (Graz, Austria) and was used as follows: 1 mg in 100 ml of methanol (2 μ l = 20 ng of pseudohypericin). (Hypericin and pseudohypericin solutions

Correspondence to: Universitäts-Professor Dr. Theodor Kartnig, Institute of Pharmakognosy, Karl-Franzens-University, Universitätsplatz 4/I, A-8010 Graz, Austria.

have to be kept in the dark and at low temperature.)

The other chemicals were as follows: pyridine (Merck, Darmstadt, Germany, product No. 9728), polyethylene glycol (PEG) 4000 (Merck, product No. 807490), viscous paraffin (Merck, product No. 7160), liquid paraffin (Merck, product No. 7162), highly liquid paraffin (Merck, product No. 7174), paraffin in block form (Merck, product No. 7150), silicone oil (Merck, product No. 7742), Triton X-100 (Merck, product No. 8603)

Thin-layer chromatography

The TLC plates were 10 × 20 cm TLC aluminium sheets coated with silica gel 60 (without fluorescent indicator) (Merck, product No. 5553) and were developed in unsaturated tanks (Desaga, Heidelberg, Germany, product No. 12 01 73). Test solutions and plant extracts were applied as bands of 6 mm (micropitette by HAACK, Vienna, Austria). The mobile phase was toluene—ethylformiate—formic acid (5:4:1) [3,4]. After development over 8 cm, solvent was quickly evaporated in a warm airstream and the plate dried for 10 min at 60°C.

Detection was with UV light at 366 nm. Hyper-

icin ($R_F = 0.58$) and pseudohypericin ($R_F = 0.63$) are separated sufficiently and with sharp boundaries.

Equipment

A Shimadzu CS 9000 dual-wavelength flyingspot scanner was used with the following settings: beam size 0.4×10 mm; wavelength = 313 nm; delta "y" = 0.04 mm, filter 4, zero set mode at start.

Sample preparation

A 30- to 50-mg aliquot of powdered (sieve 180, European Pharmacopoeia) *Hypericum* leaves or herbs was extracted with 5 ml of methanol at room temperature using a magnetic stirrer for 1 h. Extraction of plant material was repeated twice with 5 ml of methanol for 5 min. The combined filtrates were evaporated to dryness and the residue dissolved in 2.00 ml of methanol. The solution was filtered again prior to application to the TLC plate.

Preparation of the plate for fluorodensitometric determinations

The developed and dried plate was first dipped into a solution of 10% pyridine in diethyl ether (so-

TABLE I

EFFECTS OF VARIOUS SUBSTANCES ON THE INTENSITY AND STABILITY OF THE FLUORESCENCE OF HYPERICIN AND PSEUDOHYPERICIN

The stabilizing effect is expressed as the decrease in fluorescence achieved within 30 min: - = >25%; (+) = 20-25%; + = <20%; + + + = <3%.

Substance	Fold increase in intensity	Stabilizing effect	Concentration of substance; solvent
Pyridine [2,3]	2–5	-	10%; diethyl ether, acetone
PEG 4000 [7]	2-4	(+)	5%; ethanol
Pyridine + PEG 4000	2-5		5% pyridine, 5% PEG, ethanol
Highly liquid paraffin [7-11]	3-10	(+)	20-70%; diethyl ether, n-hexane, toluene
Liquid paraffin [7-11]	3-10	+	20-70%; diethyl ether, n-hexane, toluene
Viscous paraffin [7–11]	3–10	+++	20-70%; diethyl ether, n-hexane, toluene
Block form paraffin	6–8	_	4–8%; toluene
Block form + viscous paraffin	8–12	+	3-4% block form, 25-50% viscous paraffin; diethyl ether
Pyridine + highly liquid paraffin	7–10	-	1-10% pyridine, 50-70% paraffin, diethyl ether, <i>n</i> -hexane
Pyridine + liquid paraffin	7–12	-	1-10% pyridine, 50-70% paraffin, diethyl ether, <i>n</i> -hexane
Pyridine + viscous paraffin	7–15	(+)	1-10% pyridine, 50-70% paraffin, diethyl ether, <i>n</i> -hexane
Silicone oil [7]	2-10	(+)	30–70%; diethyl ether
Pyridine + silicone oil	10–12	(+)	5% pyridine, 30-70% silicone oil, diethyl ether
Triton X-100 [9-12]	2–3	(+)	1-33%; n-hexane, chloroform, toluene

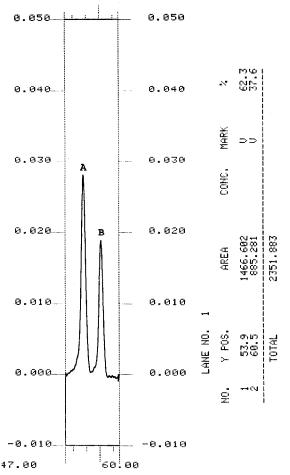


Fig. 1. Scanning profile of a methanolic extract of *Hypericum* perforatum herbs. A = Pseudohypericin; B = hypericin.

lution 1) in a dipping tank (Desaga, product No. 12 41 52). After the diethyl ether had evaporated (ca. 3 min) the plate was dipped into a homogeneous mixture of two volumes of viscous paraffin and one volume of diethyl ether (solution 2), also in a dipping tank.

Subsequently, the plate was placed vertically on a flat surface in the dark for 1 h and the emitted fluorescence was then measured.

Calculations

The contents of hypericin and pseudohypericin are calculated from the formulae:

Hypericin (ng) =
$$\frac{45 (A_1 - A_1) + 5 (A_2 - A_1)}{A_2 - A_1}$$
Pseudohypericin (ng) =
$$\frac{180 (A_1 - A_1) + 20 (A_2 - A_1)}{A_2 - A_1}$$

where A_1 is the area of the sample curve, A_1 is the area of the curve of 5 ng of hypericin and 20 ng of pseydohypericin, and A_2 is the area of the curve of 50 ng of hypericin and 200 ng of pseudohypericin.

Standards of hypericin and pseudohypericin were measured with every determination.

RESULTS AND DISCUSSION

Fluorodensitometric quantification of hypericins in situ without application of fluorescence enhancing and stabilizing reagents produces only imprecise and non-reproducible results. The reason for this phenomenon is on the one hand a relatively weak fluorescence and on the other hand insufficient stability of fluorescence intensity [7].

To optimize the quantification of hypericin and pseudohypericin we checked all the steps of the procedure. Extracting plant material in the way described dissolves out the hypericins quantitatively.

TLC separation is better achieved by the suggested system [2,3] than by using RP-8 plates and acetonitrile [1]. The spots are separated well and exhibit sharp margins, as the scanning profile shows (Fig. 1).

To increase fluorescence intensity and improve stability, we tested several substances in various concentrations and combinations as well as in various solvents (see Table I).

Homogeneous application of the reagents to the plate is only guaranteed by dipping and not by spraying [7]. The best effect was achieved with 10% pyridine in diethyl ether as the first dipping solution. After removing diethyl ether from the plate in a warm airstream, other solutions were applied as second reagents (see Table II).

Finally, we found the best effect using consecutive applications of 10% pyridine in diethyl ether as the first dipping solution and 66% viscous paraffin in diethyl ether as the second dipping solution.

To conserve solvents, the dipping procedure should be carried out in dipping tanks. The TLC plate has to be held with forceps. Immediately after withdrawing the plate from the second solution, it

TABLE II
EFFECTS OF VARIOUS SUBSTANCES APPLIED AFTER A BASIC TREATMENT WITH 10% PYRIDINE IN METHANOL

The stabilizing effect is expressed as the decrease in fluorescence achieved within 30 min. - = >25%; (+) = 20-25%; + = <20%; + + + = <3%.

Substance	Fold increase in intensity	Stabilizing effect	Concentration of substance; solvent
PEG 4000	2		5%; ethanol
Liquid paraffin	12-14	+	70%; diethyl ether
Viscous paraffin	6–15	+++	50-70%; diethyl ether
Blijock form paraffin	6–8	_	4-8%; toluene
Bock form + viscous paraffin	10–12	-	3-4% block form, 25-50% viscous paraffing diethyl ether
Silicone oil	10		70%; diethyl ether
Triton X-100	2–3	_	1-33%; n-hexane, chloroform, toluene

should be placed vertically on a flat surface, e.g. on the turned over metal cover of a TLC tank (Desaga, product No. 12 01 02), in the dark for 1 h. During this time the diethyl ether evaporates and the excess paraffin flows down the plate. After 1 h fluorescence is sufficiently stable for measurements to be made.

To maintain exact results with every TLC plate, 5 and 50 ng of hypericin as well as 20 and 200 ng of pseudohypericin have to be determined as calibration substances. In numerous determinations of identical samples we observed that, in spite of strict adherence to the protocol, there were small differences from experiment to experiment. These differences can be avoided by using hypericin and pseudohypericin standards on the same plate.

The accuracy of the fluorodensitomeric measurement of hypericin and pseudohypericin in the procedure described is expressed as the relative standard deviation and is less than 1.5%. The reproducibility of the determination of hypericin and pseudohypericin from plant material is also expressed as the relative standard deviation and is less than 3.5%.

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REFERENCES

- M. Vanhaelen and R. Vanhaelen-Fastre, J. Chromatogr., 281 (1983) 263.
- 2 H. Wagner, S. Bladt, and E.-M. Zgainski, Drogenanalyse, Dünnschichtchromatographische Analyse von Arzneidrogen, Springer, Berlin, 1983.
- 3 Deutscher Arzneimittel-Codex (Ergänzung zum Arzneibuch), Govi-Verlag, 1979.
- 4 R. Berghöfer and J. Hölzl, *Dtsch. Apoth. Ztg.*, 126 (1986) 2569.
- 5 P. Pachaly, Dtsch. Apoth. Ztg., 124 (1984) 2159.
- 6 C. F. Poole, S. K. Poole, T. A. Dean and N. M. Chirco, J. Planar Chromatogr., 2 (1989) 180.
- 7 H. Jork, W. Funk, W. Fischer and H. Wimmer, Dünnschichtchromatographie. Reagenzien und Nachweismethoden, Band 1a, VCH Verlagsgesellschaft 1989.
- 8 S. Uchiyama and M. Uchiyama, J. Chromatogr., 153 (1978)
- 9 S. Uchiyama and M. Uchiyama, J. Chromatogr., 262 (1983)
- 10 B. Lin Ling, W. R. G. Baeyens, H. Marysael, K. Stragier and P. de Moerloose, J. Liq. Chromatogr., 12 (1989) 3135.
- 11 S. Uchiyama and M. Uchiyama, J. Liq. Chromatogr., 3 (1980) 681.
- 12 S. S. J. Ho, H. T. Butler and C. F. Poole, J. Chromatogr., 281 (1983) 330.